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RC Schank

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alcohols ethanol 2 methylpropan 2 ol butan 2 ol but 2 en 1 ol
 ch₃ch₂ch₂ch₂oh ii ch₃ ch₃ch₂ch₂oh iii ch₃
 ch₃ch₂ch₂oh iv ch₃ ch₃cch₂oh ch₃ 6 aldehydes and
 ketones draw the structures for the following compounds
 ethanal butanone also known as butan 2 one but the 2 is
 unnecessary and is often left out 2 methylpropanal
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question 1 basic hydrocarbon nomenclature part a enter the

number of carbons indicated by each of the following root
 names in the designated answer box part b the line formula
 for a branched alkane is shown below part c the line formula
 for another branched alkane is

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jan 21 2020 the point of attachment is number one and any
 other smaller groups are named as substituent groups on the
 larger group this numbering is independent of the
 numbering of the parent chain try to name the following
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 really have got this one wrong unless you made a careless
 mistake like putting too many or too few hydrogens on the
 middle carbon if you got it wrong re read the bit about

naming alkanes and then look at these questions again ii
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dec 16 2021 so the complete name is 5 bromo 7 chloro 6 hydroxy 2 2 5 trimethyl 7 octen 4 one 4 it is not difficult to find the parent structure for this compound which is a cyclic alcohol so the last name is cyclopropanol the naming of the substituent with the benzene ring is

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solutions 1 aromatic compound iodobenzene 2 alkane 3 methylhexane 3 ketone 4 alkane alkyl halide 3 chloro 4 ethyl 2 4 dimethyloctane 5 aromatic compound o diethylbenzene or ortho diethylbenzene 6 alkene 3 methylpentene 7 ester 8 aromatic compound p bromotoluene or para bromotoluene 9 alkyne

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feb 21 2017 give the complete iupac name for each of the following compounds a b c give the complete iupac name and the common name for each of the following compounds a b c a 1 ethoxybutane butyl ethyl ether b 1 ethoxypropane ethyl propyl ether c 1 methoxypropane methyl propyl ether

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cyclichydrocarbons

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1 carbon methane 2 carbons ethane 3 carbons propane 4 carbons butane 5 carbons pentane 6 carbons hexane 7 carbons heptane 8 carbons octane 9 carbons nonane 10 carbons decane while many of these names share a greek root with

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